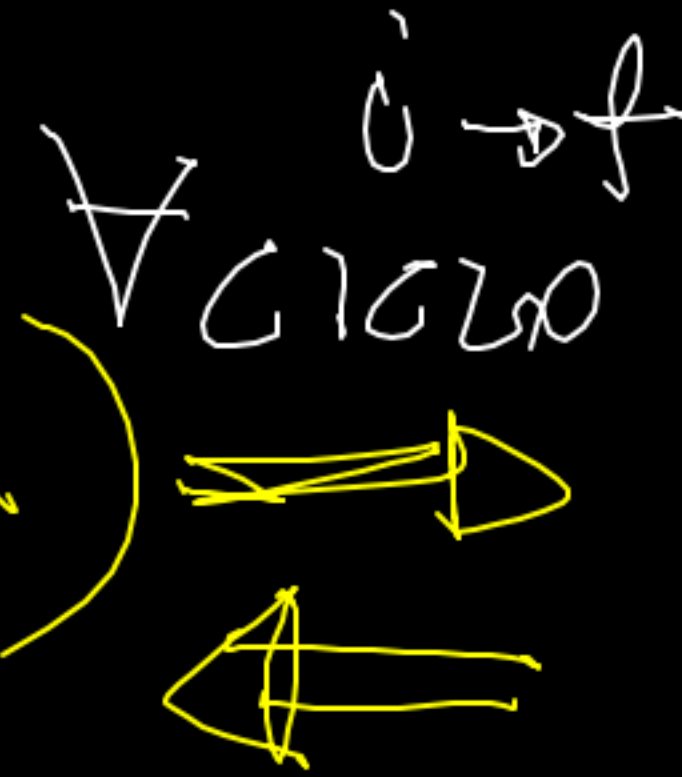
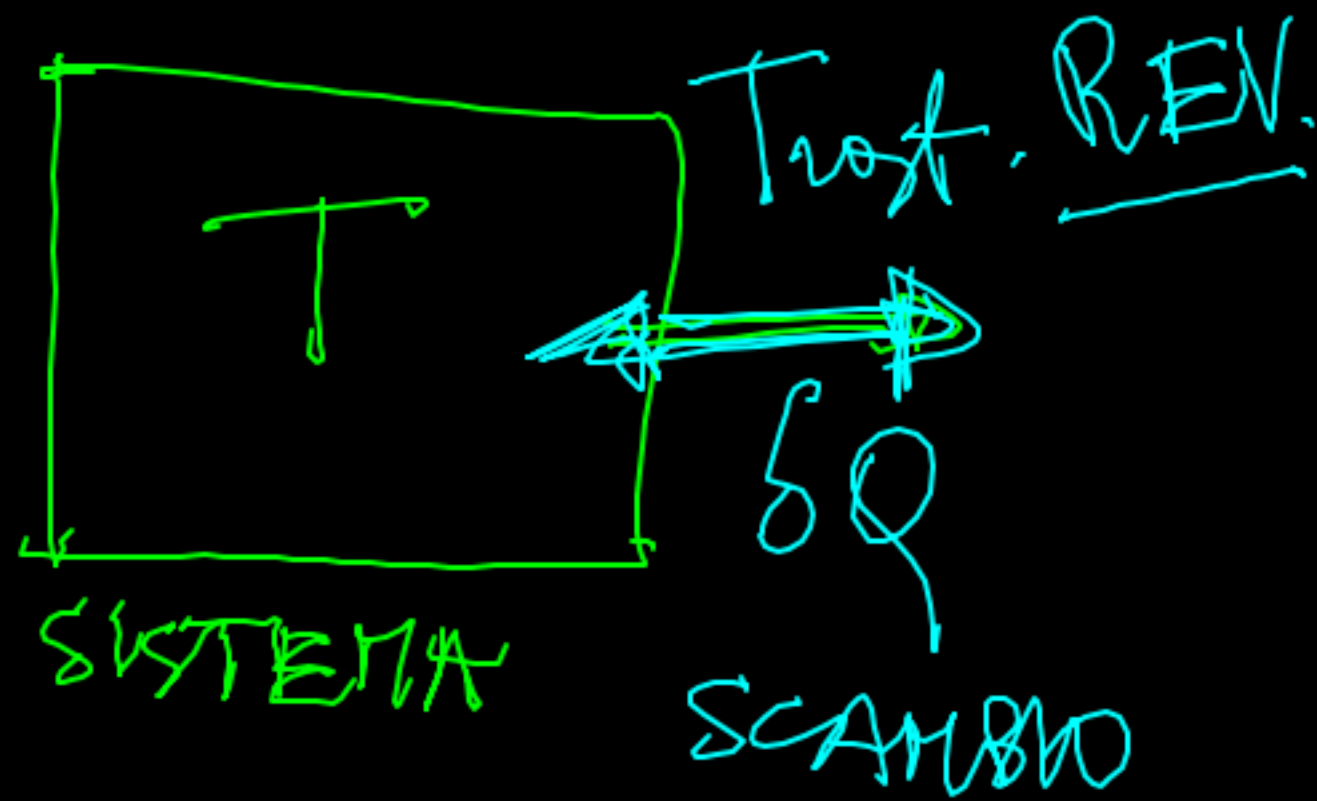
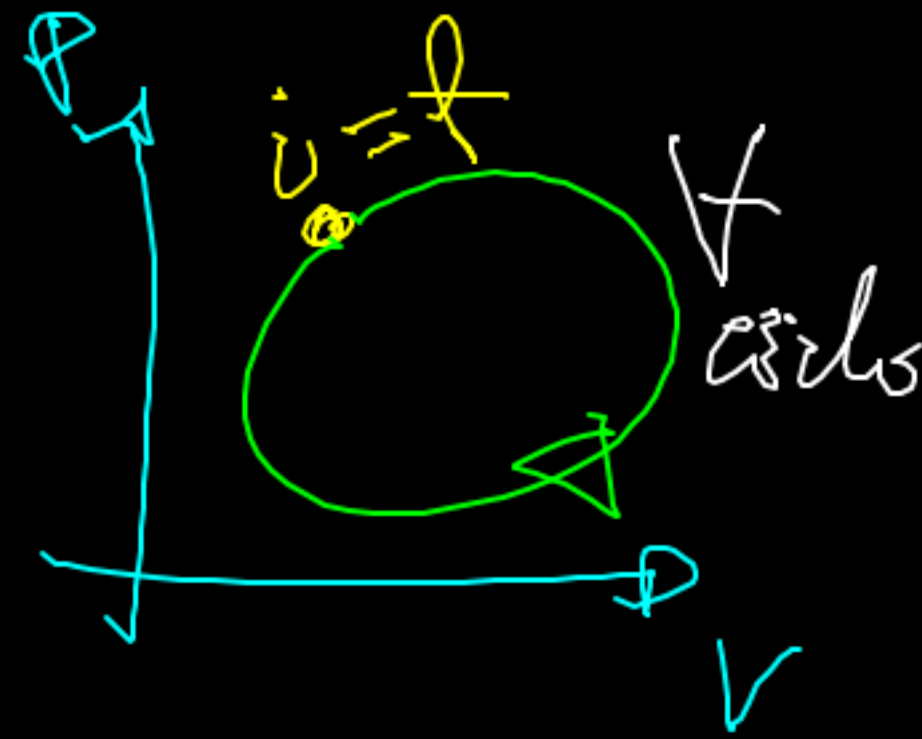


# ENTROPIA

→ Variabili di Stato ( $P, V, T, U, \dots$ )



$$\begin{aligned} \Delta P &= 0 \\ \Delta U &= 0 \\ \Delta T &= 0 \\ \Delta V &= 0 \\ &\vdots \end{aligned}$$



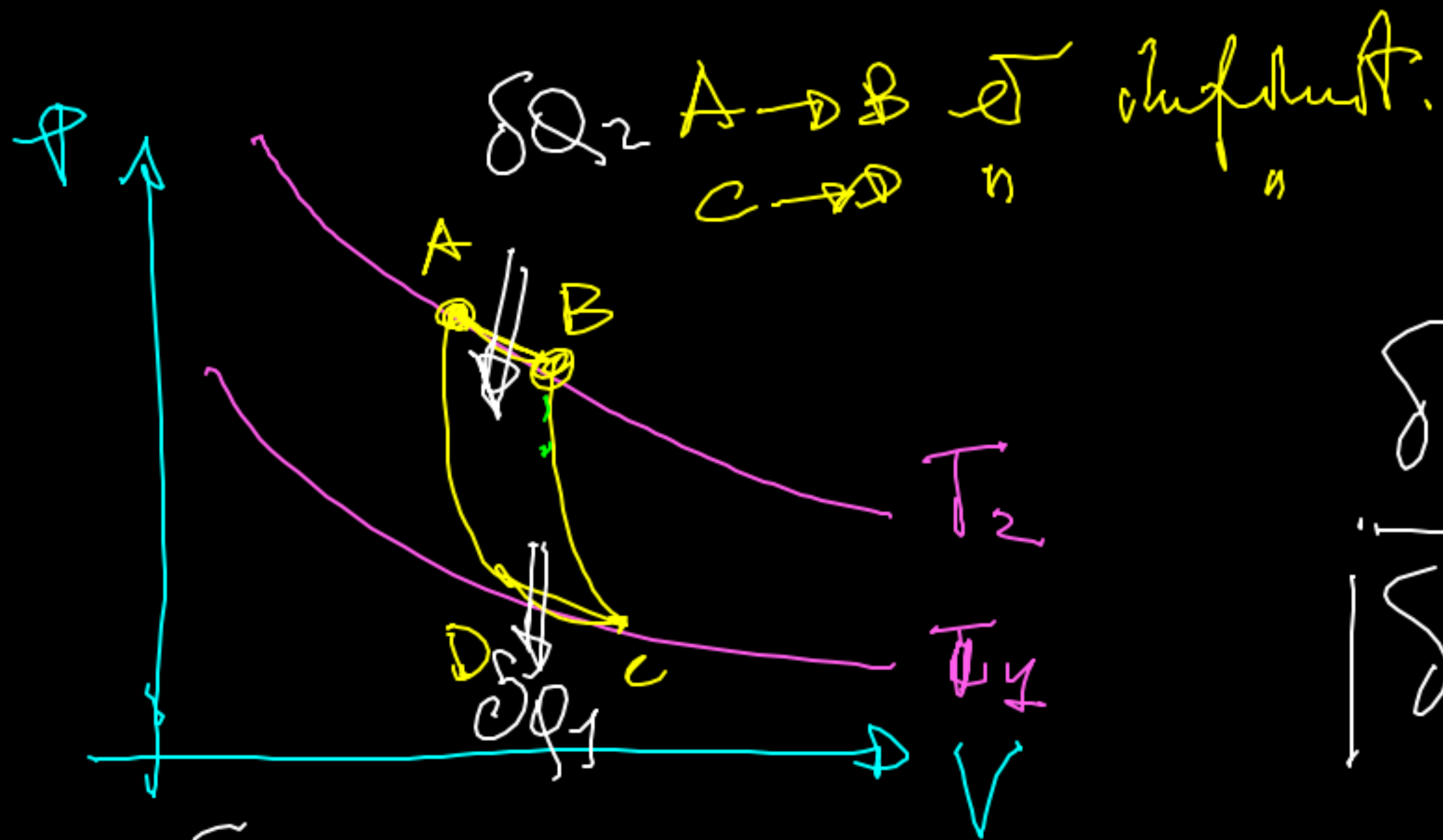
$$dS = \frac{\delta Q}{T}$$

VARIAZIONE DI ENTROPIA

$\Delta S = 0$   
 SU UN  
 CICLO?

PER UNA TRASF. FINITA REV.

$$\Delta S = \int_i^f \frac{\delta Q}{T} = S_f - S_i$$



$\delta Q_4 < 0$   
 $\delta Q_2 > 0$

$dS_{AD} = 0$

per ciclo  $\delta Q = 0$

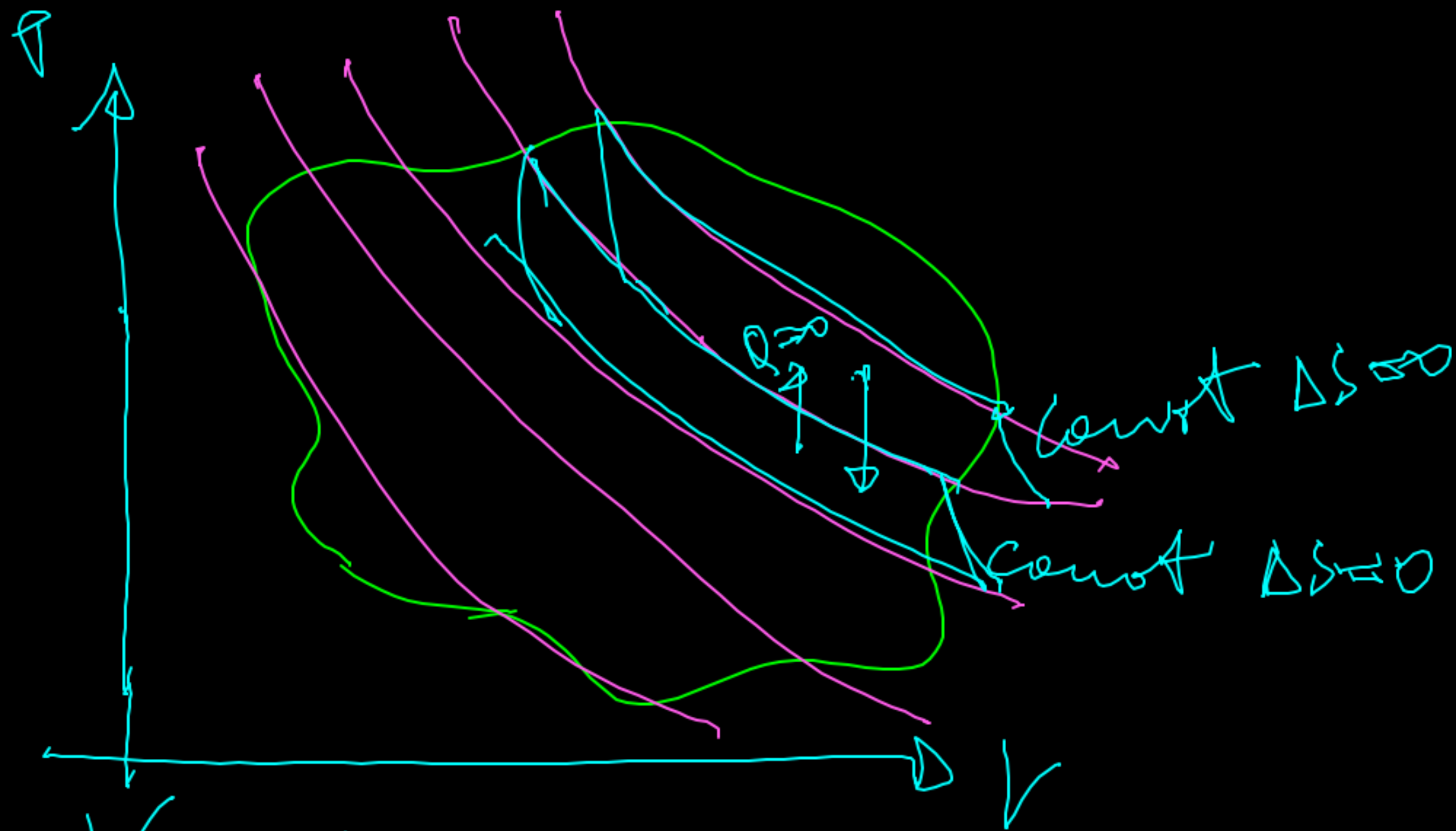
$$\frac{\delta Q_2}{|\delta Q_1|} = \frac{T_2}{T_1} \Rightarrow \frac{\delta Q_2}{T_2} = \frac{|\delta Q_1|}{T_1}$$

$$\frac{\delta Q_2}{T_2} - \frac{|\delta Q_1|}{T_1} = 0 = \frac{\delta Q_2}{T_2} + \frac{\delta Q_1}{T_1}$$

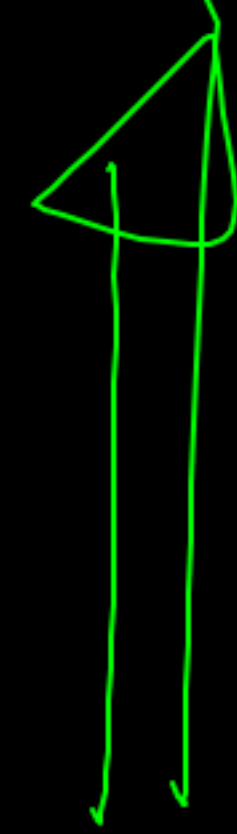
PER UN CICLO  
DI CARNOT REV.

$$0 = dS_2 + dS_1$$

$\Delta S = 0$



L'ENTROPIA  
 È UNA FUNZIONE  
 DI STATO



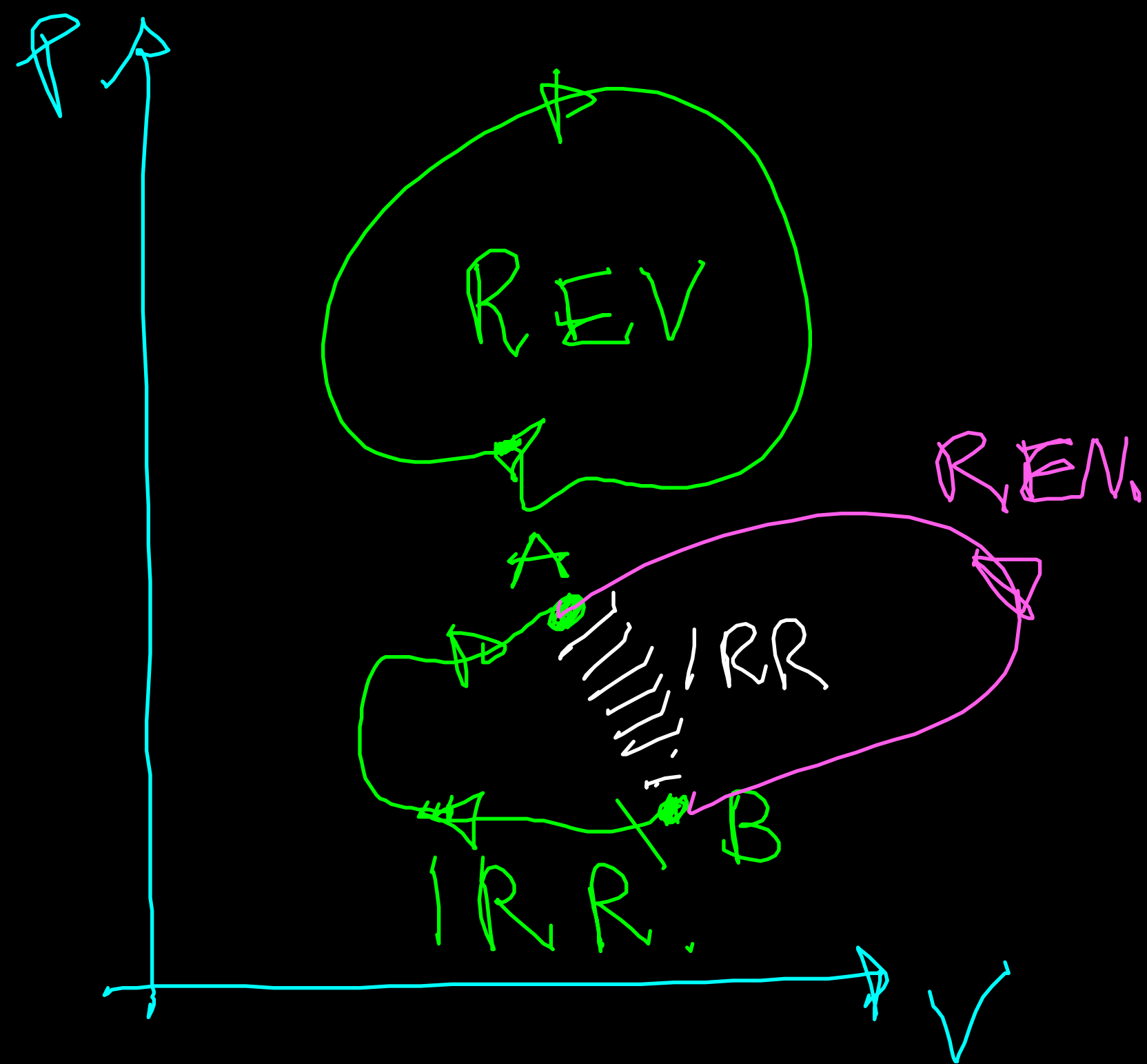
$\nabla$  CICLO REV.  $\rightarrow$  SOMMA DI CICLI DI CARNOT REV.  
 $\Delta S = 0 \rightarrow \Delta S = 0$

$$\Delta S_{\text{ciclo}} = 0 \quad (\text{ciclo qualsiasi})$$

$$\Delta S_{\text{ciclo}} = \int_{\text{CICLO REV.}} \frac{\delta Q}{T} = \oint_{\text{REV.}} \frac{\delta Q}{T} = 0$$

$S$  è una  
funzione  
di stato

Variabili di Stato:  $p, V, n, T, U, S, \dots$



$A \rightarrow B$   
 TRASF.  
 REALE IRR.

si può comunque calcolare  $\Delta S_{AB} = S_B - S_A$   
 usando una qualunque  
 trasformazione REV. da  $A \rightarrow B$

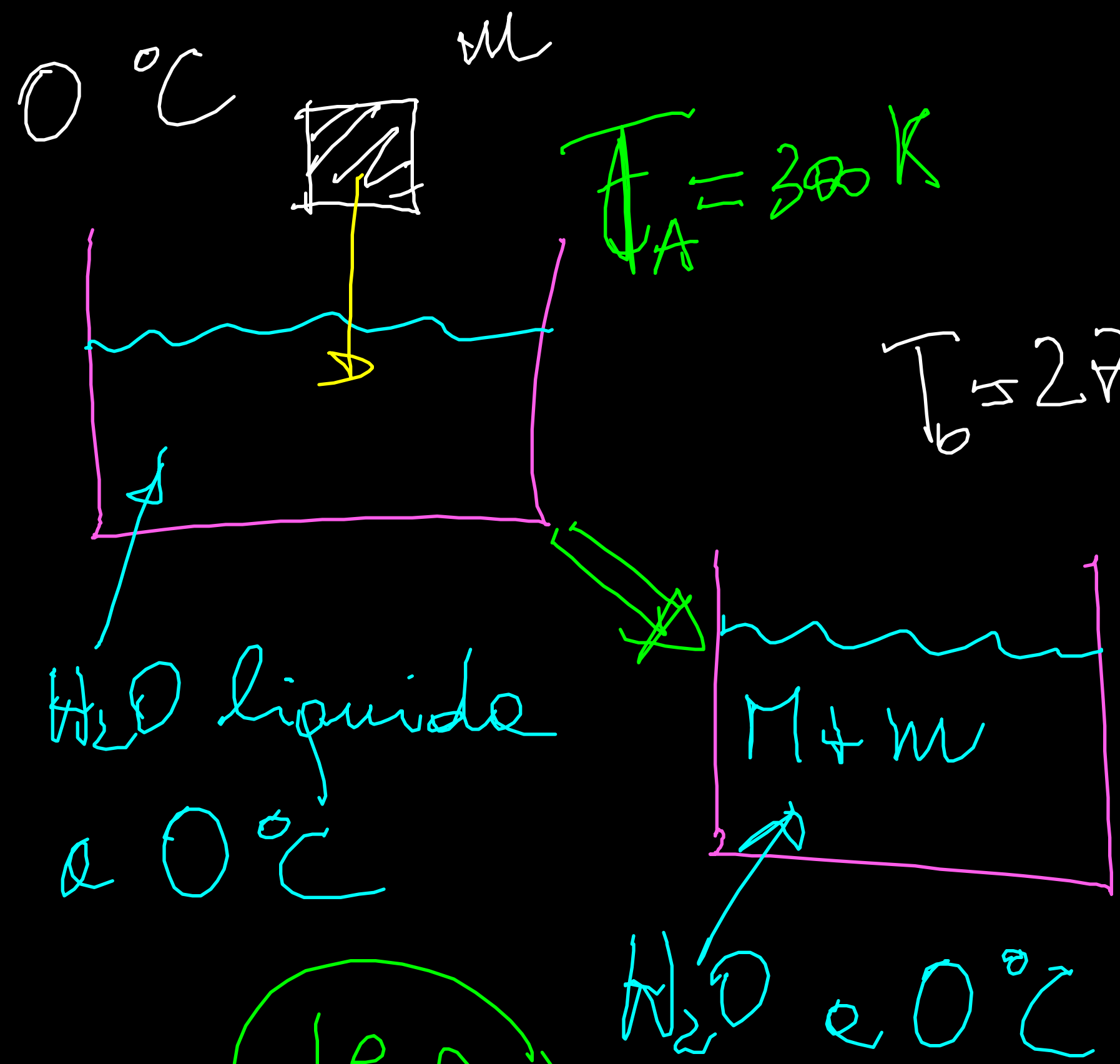
$$\oint \delta Q \begin{cases} = 0 & \text{REV.} \\ < 0 & \text{IRR.} \end{cases}$$

INTEGRALE DI CLAUSIUS



# ESEMPDI DI CALCOLO DI $\Delta S$

• Cambiamento di fase

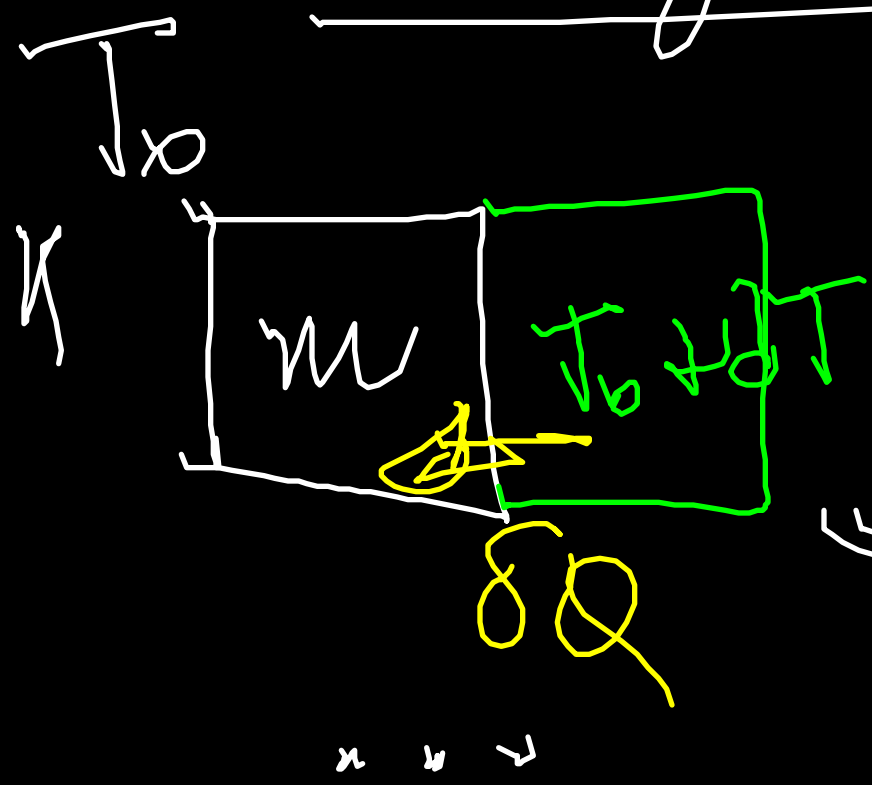


IRR

Per calcolare  $\Delta S_{\text{fusione}}$

immagina una Tratt. rev.

INFINITE QUASI-STATICHEREV.

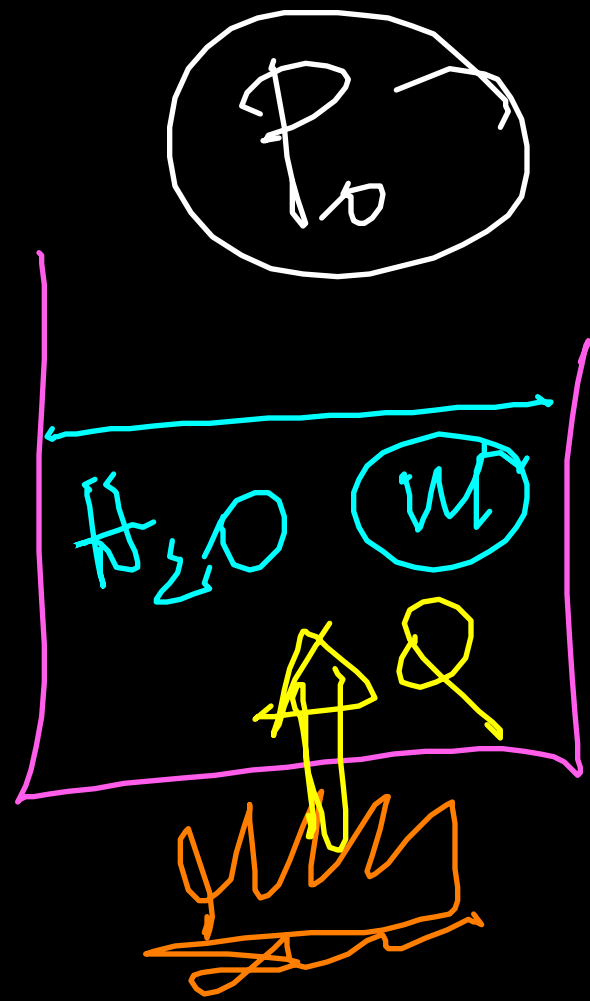


$$Q_{\text{tot}} = m \lambda_f$$

$$\Delta S_{\text{fusione}} = S_{\text{Liq}} - S_{\text{Solido}} = \int \frac{\delta Q}{T} = \frac{1}{T_0} \int \delta Q = \frac{Q_{\text{tot}}}{T_0}$$

$$\Delta S_{\text{fusione}} = \frac{Q_{\text{tot}}}{T_0} = \frac{m \lambda_f}{T_0} > 0 \quad \left[ \frac{J}{K} \right]$$

• riscaldamento



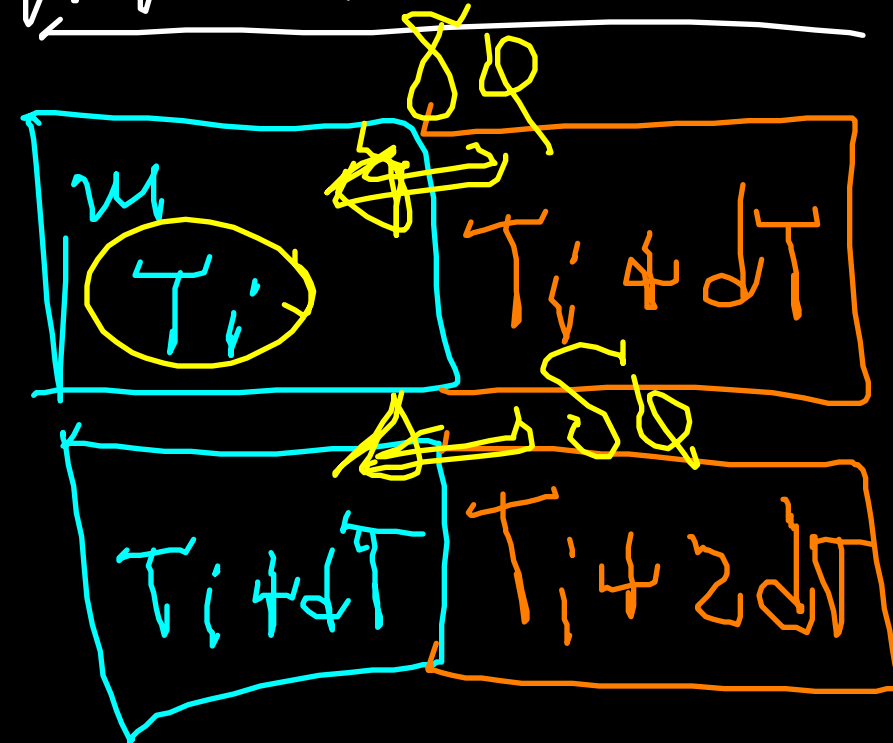
$H_2O$

$t_i \rightarrow t_f < 100^\circ C$

$T_i \rightarrow T_f < 373.16 K$

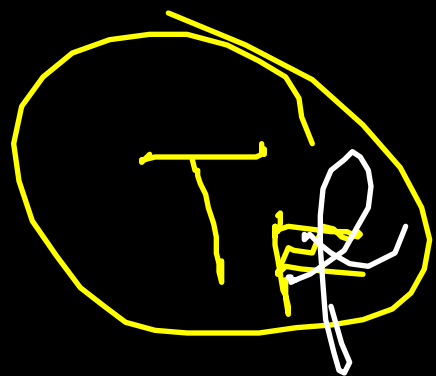
IRR.

MMAGNO



quasi-static  
rev.

delta Q



$$\Delta S = \int_{T_i}^{T_f} \frac{\delta Q}{T} = \int_{T_i}^{T_f} \frac{m C_p dT}{T}$$

$$= m C_p \int_{T_i}^{T_f} \frac{dT}{T} = m C_p \ln \frac{T_f}{T_i} > 0$$

$T_f > T_i$