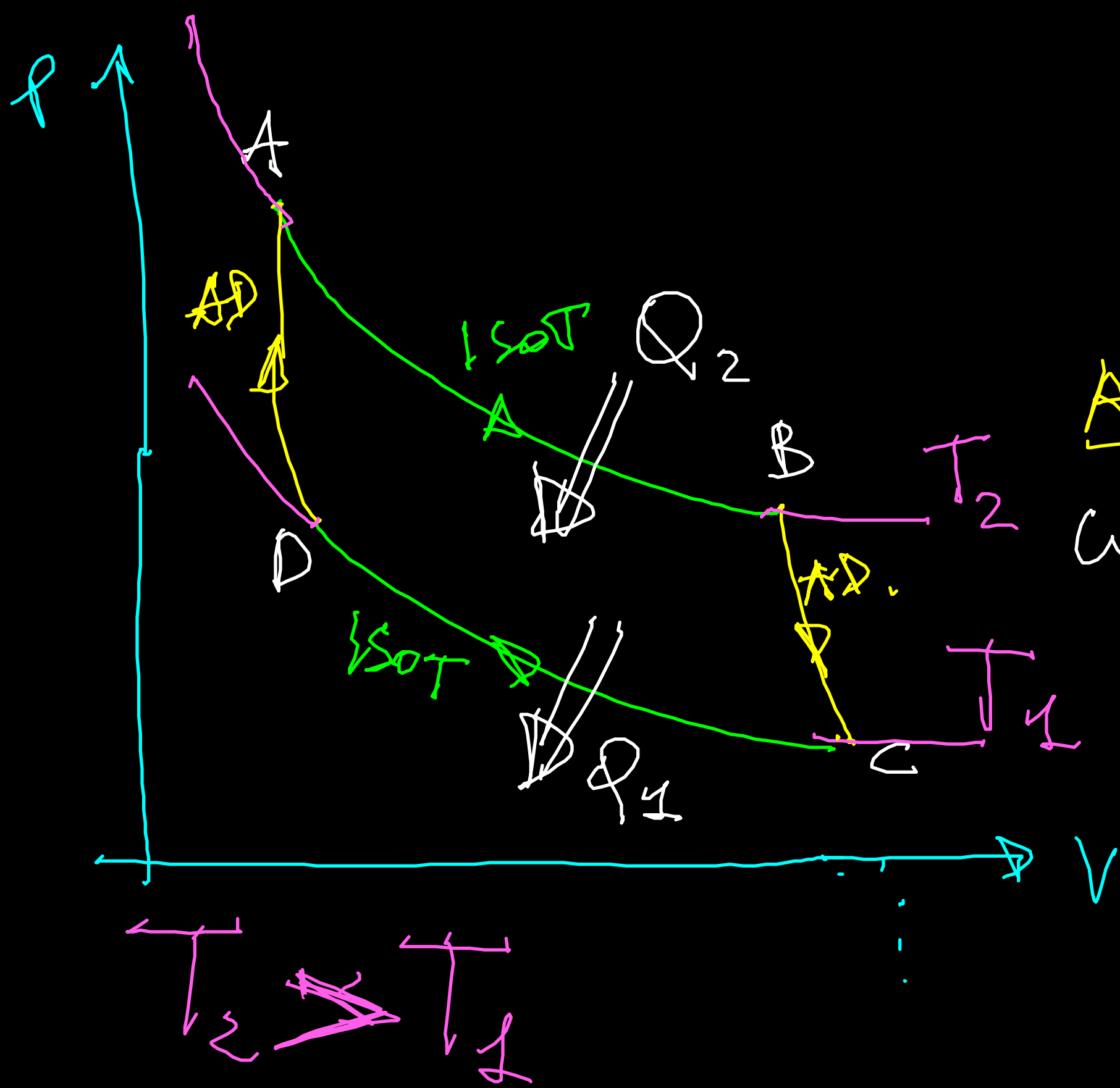


CICLO DI CARNOT



$$\eta = 1 - \frac{n R T_1 \ln \frac{V_C}{V_D}}{n R T_2 \ln \frac{V_B}{V_A}} = 1 - \frac{T_1}{T_2}$$

ADIAB

$$const = p V^\gamma = p V V^{\gamma-1} =$$

$$= n R T V^{\gamma-1} \Rightarrow p T V^{\gamma-1} = const$$

B → C

$$T_1 V_C^{\gamma-1} = T_2 V_B^{\gamma-1}$$

D → A

$$T_1 V_D^{\gamma-1} = T_2 V_A^{\gamma-1}$$

rapporto

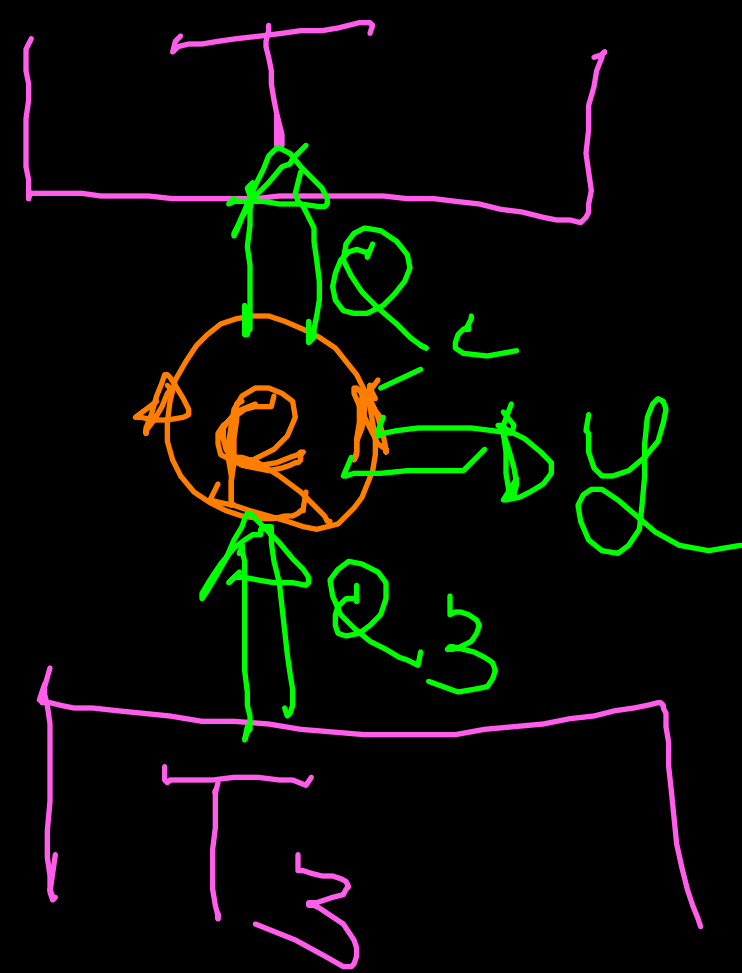
$$\left(\frac{V_C}{V_D}\right)^{\gamma-1} = \left(\frac{V_B}{V_A}\right)^{\gamma-1} \Rightarrow \frac{V_C}{V_D} = \frac{V_B}{V_A}$$

$$\eta = 1 - \frac{|Q_1|}{Q_2}$$

Rendimento di un ciclo di Carnot (tra due sole sorgenti)

$$\eta_c = 1 - \frac{T_c}{T_F} > \eta_{\text{max}}$$

Temperatura Termodinamica



$$\eta_c = 1 - \frac{|Q_3|}{Q_c} = 1 - \frac{T_3}{T} = \eta_R$$

$$\frac{T}{T_3} = \frac{Q_c}{|Q_3|} \Rightarrow$$

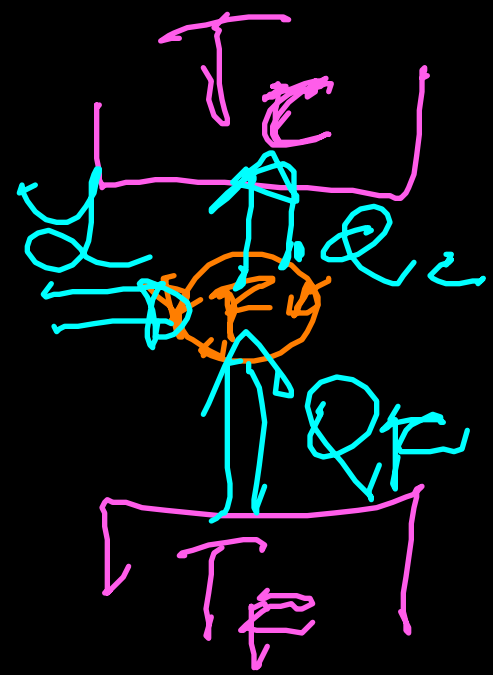
$T_3 \equiv$ punto triplo dell'acqua $\rightarrow 273.16 \text{ K}$

$$T = (273.16) \frac{Q_c}{|Q_3|}$$

TEMP. TERMODIN. ASSOLUTA

Per una macchina frigorifera rev.

$$K = \frac{Q_F}{|Q_c|} = \frac{Q_F}{|Q_c| - Q_F} = \frac{T_c}{T_c - T_F}$$

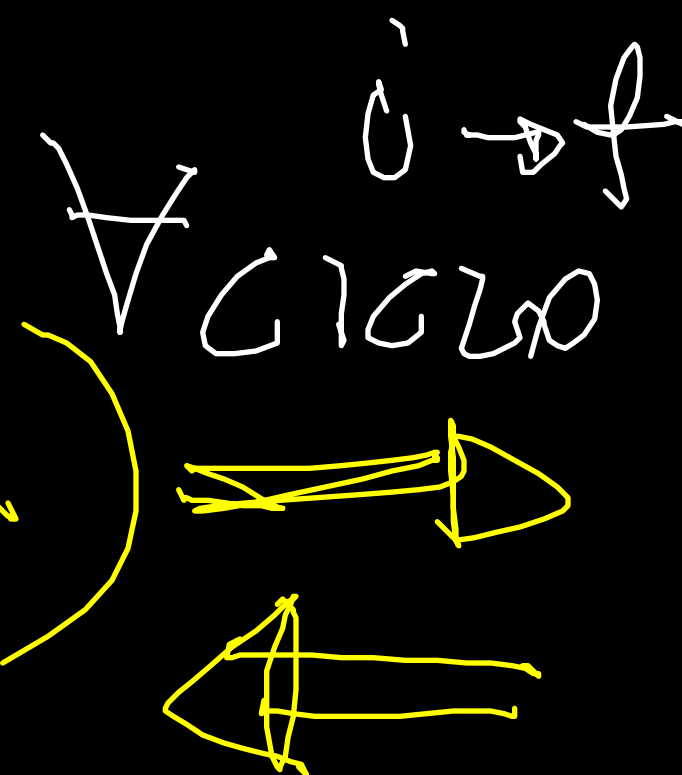


$$\frac{T_F}{T_c} = \frac{Q_F}{|Q_c|}$$

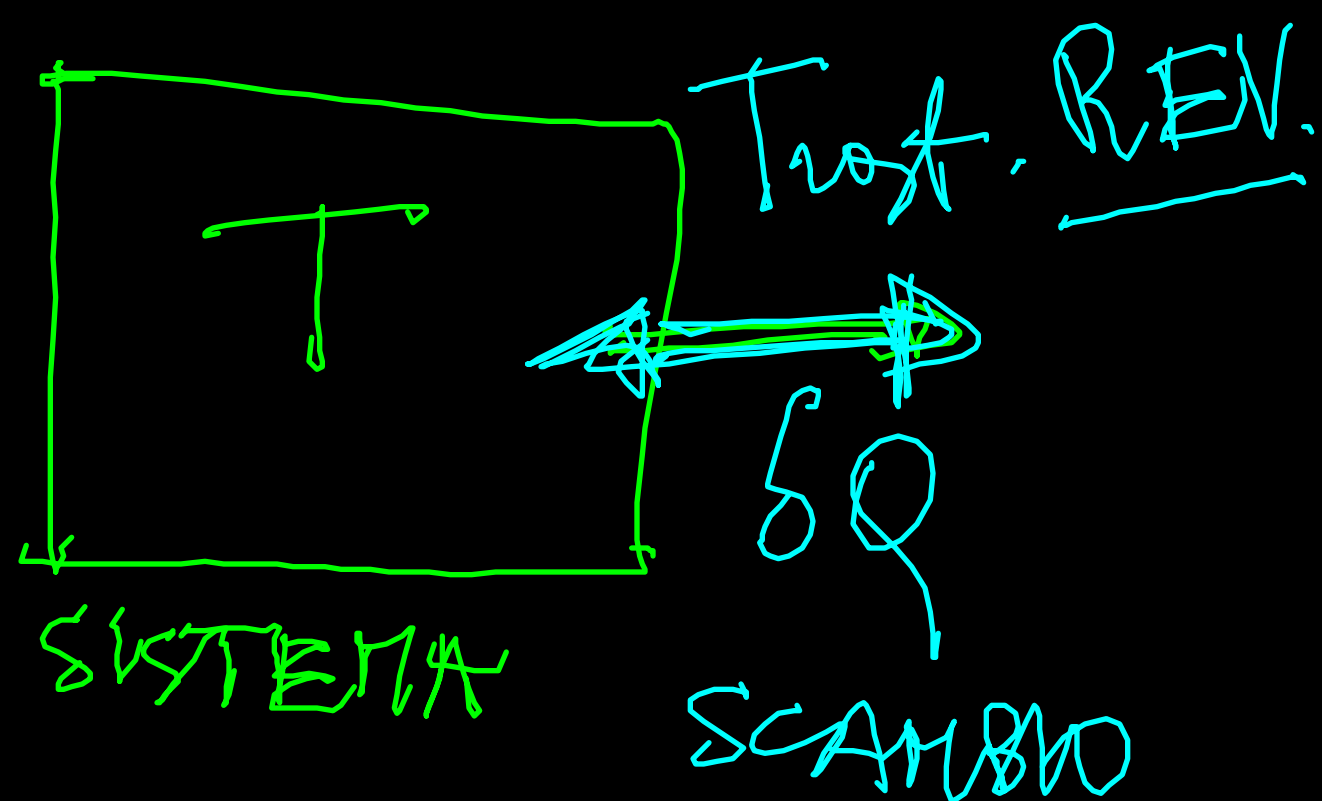
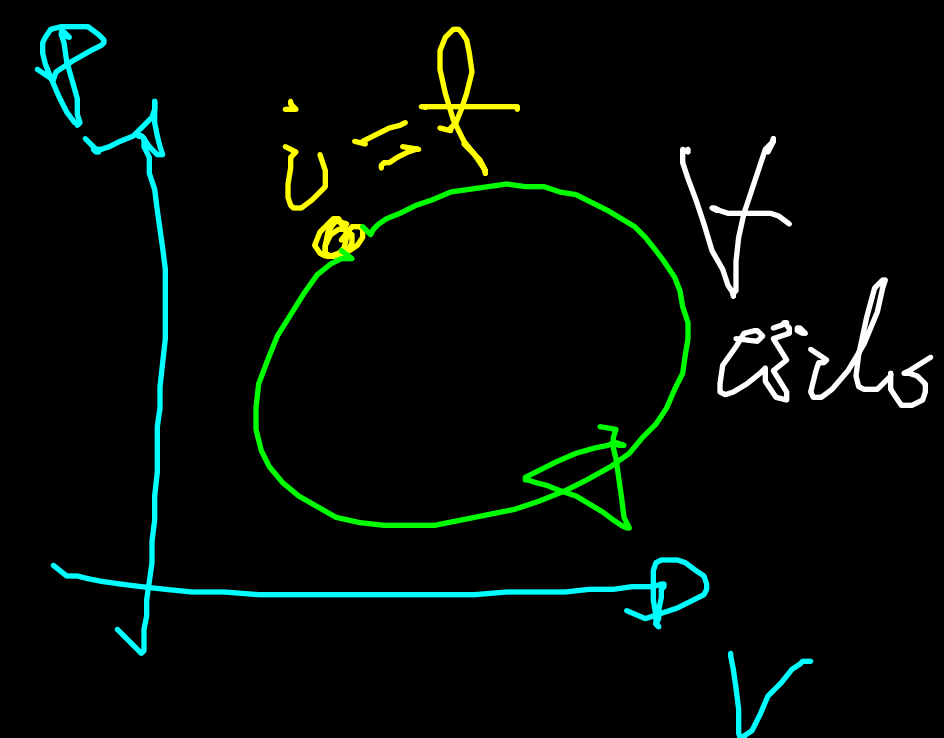
Coeff. Di
Prestazione
massimo
possibile

ENTROPIA

→ Variabili di Stato (P, V, T, U, \dots)



$$\begin{aligned} \Delta P &= 0 \\ \Delta U &= 0 \\ \Delta T &= 0 \\ \Delta V &= 0 \end{aligned}$$



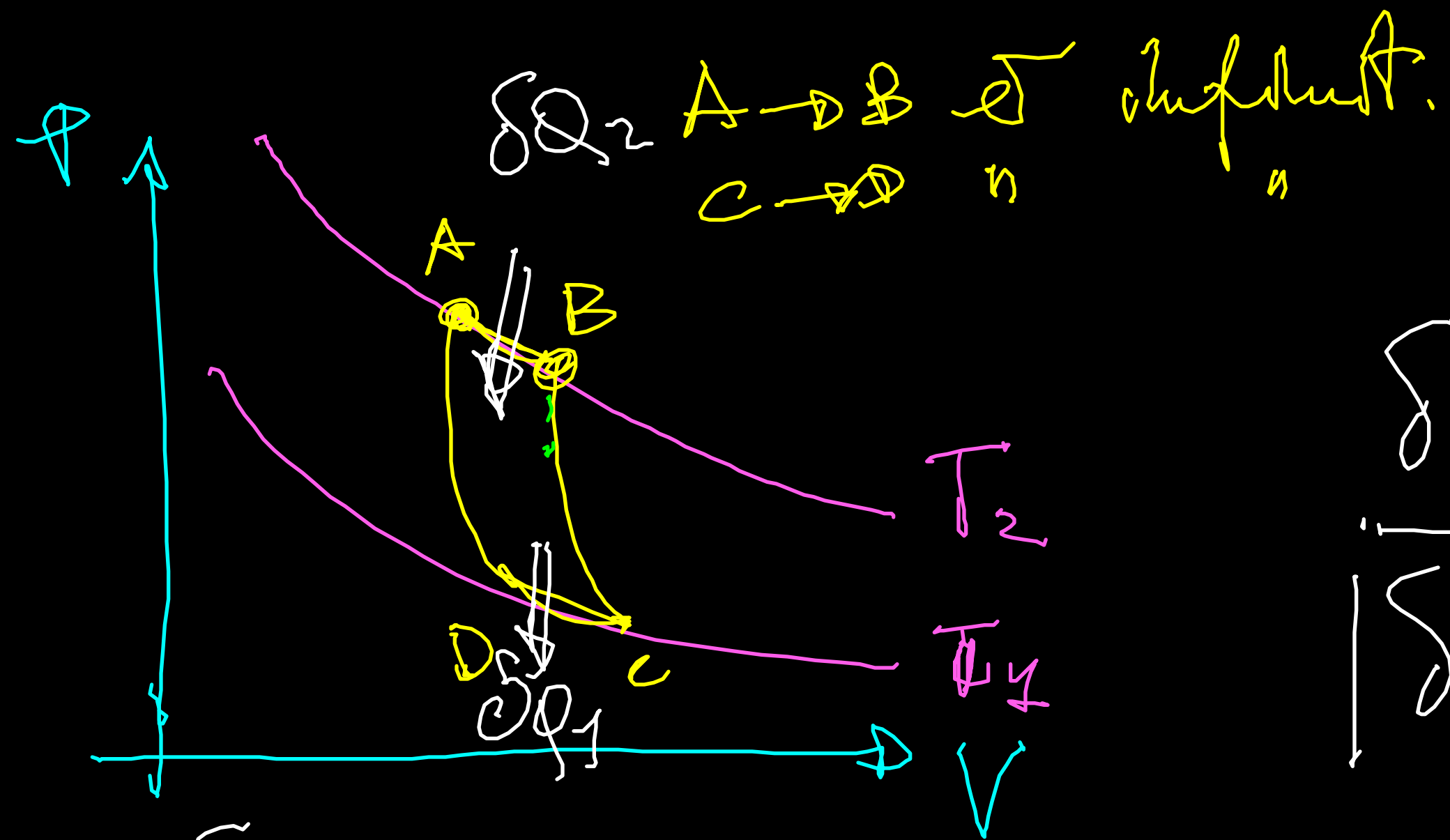
$$dS = \frac{\delta Q}{T}$$

VARIAZIONE DI ENTROPIA

$\Delta S = 0$
 SU UN
 CICLO?

PER UNA TRASF. FINITA REV.

$$\Delta S = \int_i^f \frac{\delta Q}{T} = S_f - S_i$$



$$\frac{\delta Q_2}{|\delta Q_1|} = \frac{T_2}{T_1} \Rightarrow \frac{\delta Q_2}{T_2} = \frac{|\delta Q_1|}{T_1}$$

$$\delta Q_4 < 0$$

$$\delta Q_2 > 0$$

$$dS_{AD} = 0$$

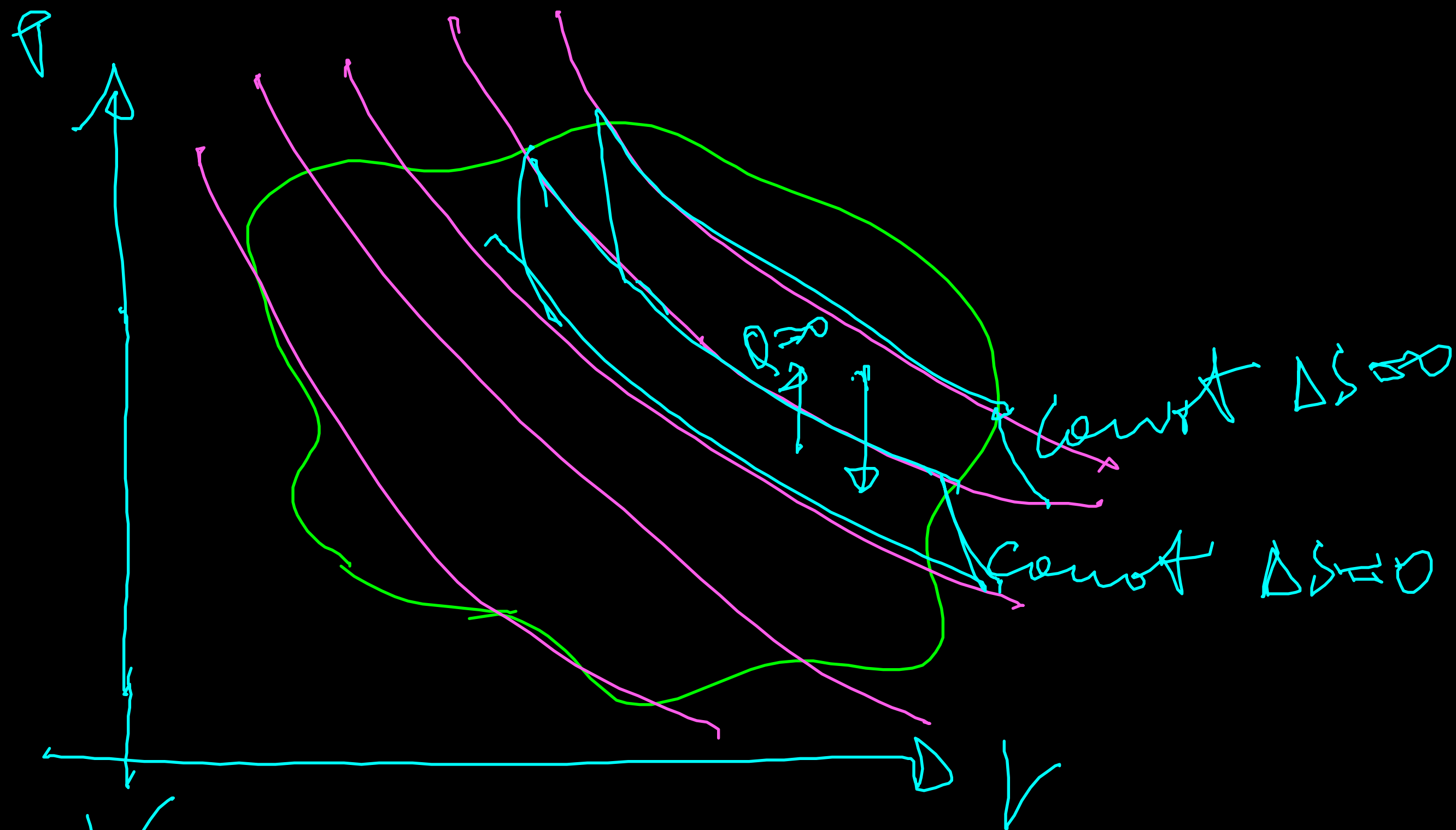
$$\text{per } \delta Q = 0$$

$$\frac{\delta Q_2}{T_2} - \frac{|\delta Q_1|}{T_1} = 0 = \frac{\delta Q_2}{T_2} + \frac{\delta Q_1}{T_1}$$

PER UN CICLO
DI CARNOT

$$0 = dS_2 + dS_1$$

$$\Delta S = 0$$



L'ENTROPIA
 È UNA FUNZIONE
 DI STATO

